Endothermic reactions

Reactions in which energy is absorbed are called endothermic reactions.

29:

Fe Sog. 7H20 - Fe Sog +7H20

2 Fe SO Fe 20 + SO + SO 3

Ca(03(5) - Ca(s) + (02(3)

2P6(NO3) 2 P60(s) + 4NO2(g) + O2(g)

 $2H_2O(2)$ electricity $2H_2(g) + O_2(g)$

2 Agu (s) sunlight, 2 Ag (s) + c/2(g)

aAg Br (s) sunlight aAg(s) + Br2(g)

Ba(OH) 2. 8 H2 O (s) + 2NH2 ((s)

The temperature after this reaction drops from room temperature to - 220. This is an endothermic reaction.