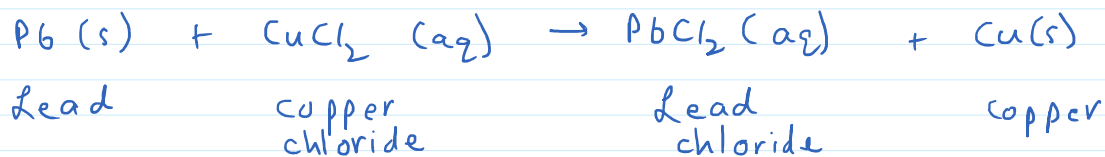
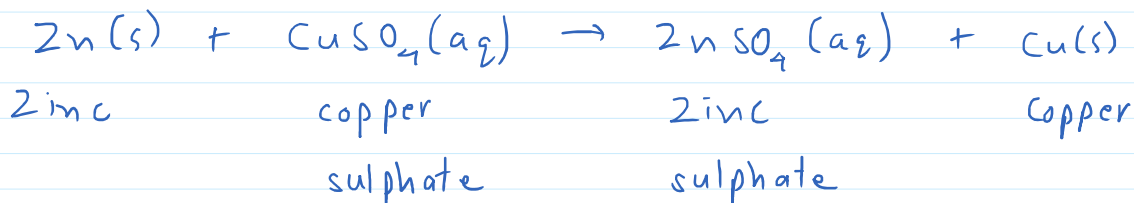
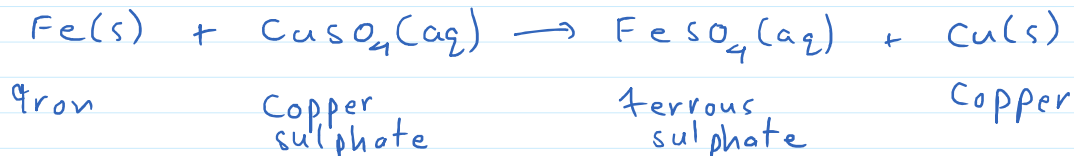


Displacement reaction:

A reaction in which more reactive element displaces less reactive element from a compound is called displacement reaction.

eg:



Elements which are more reactive displace elements which are less reactive from their compounds.

Fe, Pb and Zn are more reactive than Cu.

Double displacement reaction

Reactions in which there is exchange of ions between the reactants are called double displacement reactions.

eg:

